## Spring 2009 CH302 Practice Quiz 6

1. Consider the balanced reaction below:

2X (s) + 2H<sub>2</sub>O (l)  $\rightarrow$  2XOH (aq) + H<sub>2</sub> (g)

The species 'X' would be which of the following?

- 1. a halogen
- 2. an alkaline earth metal
- 3. an alkali metal
- 4. a chalcogen
- 2. Which of the following is not true of alkaline earth metals?
  - 1. React with halogens to form salts
  - 2. Tend to form a +2 charge
  - 3. Somewhat reactive toward water
  - 4. Gain electrons to achieve noble gas configuration
  - 5. Have 2 electrons in their highest energy shell
- 3. Which member of the Boron family is a deadly poison
  - 1. Aluminum (Al)
  - 2. Boron (B)
  - 3. Gallium (Ga)
  - 4. Indium (In)
  - 5. Thallium (TI)

4. The Nitrogen group contains (1/2) non-metal(s), (1/2) metalloid(s) and (1/2) metal(s). Do not consider the synthetic superheavy element Ununpentium (Uup) in arriving at your answer.

- 1.2,1,2
- 2.1,2,2
- 3. 2, 2, 1
- 4. 1, 1, 1
- 5.2,2,2

5. Which member of the carbon family is most abundant in Earth's crust?

- 1. Silicon (Si)
- 2. Carbon (C)
- 3. Germanium (Ge)
- 4. Tin (Sn)
- 5. Lead (Pb)
- 6. Which of the following statements is not true of the oxygen family?
  - 1. Are good reducing agents
  - 2. They often have an oxidation number of -2
  - 3. Contains elements crucial to life
  - 4. Are also called chalcogens
- 7. Alumina (Al<sub>2</sub>O<sub>3</sub>) is produced in which of the following processes?
  - 1. Hall process
  - 2. Contact process
  - 3. Bayer process
  - 4. Claus process

8. Which of the following gemstones is/are derived from aluminum oxides?

- I. Diamond
- II. Sapphire
- III. Ruby
- 1. I
- 2. II
- 3. III
- 4. I and II
- 5. I and III

6. II and III 7. none